

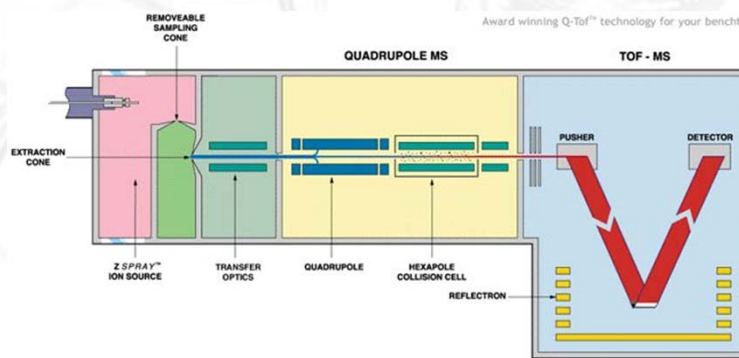
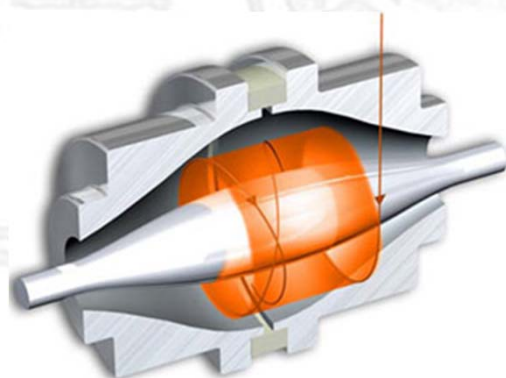
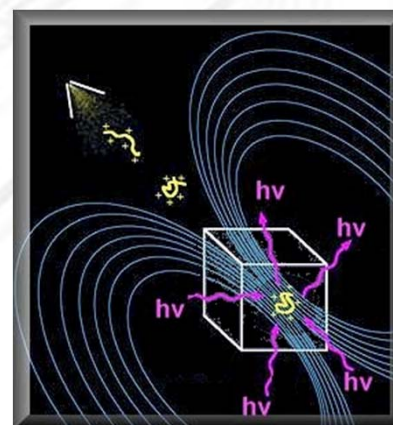
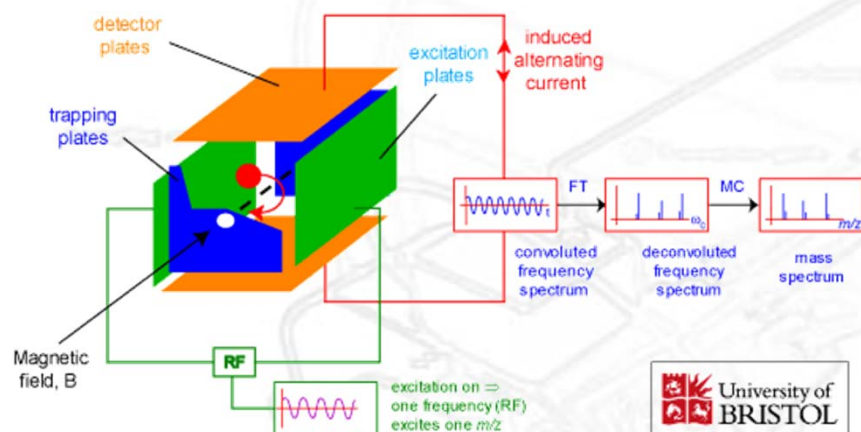
## Week 6: Vacuum Systems







































COOLVAC 2.000 CL similar

# Last Time...

- Fourier Transform Mass Analyzers, Hybrid Instruments, Other Dissociation Methods



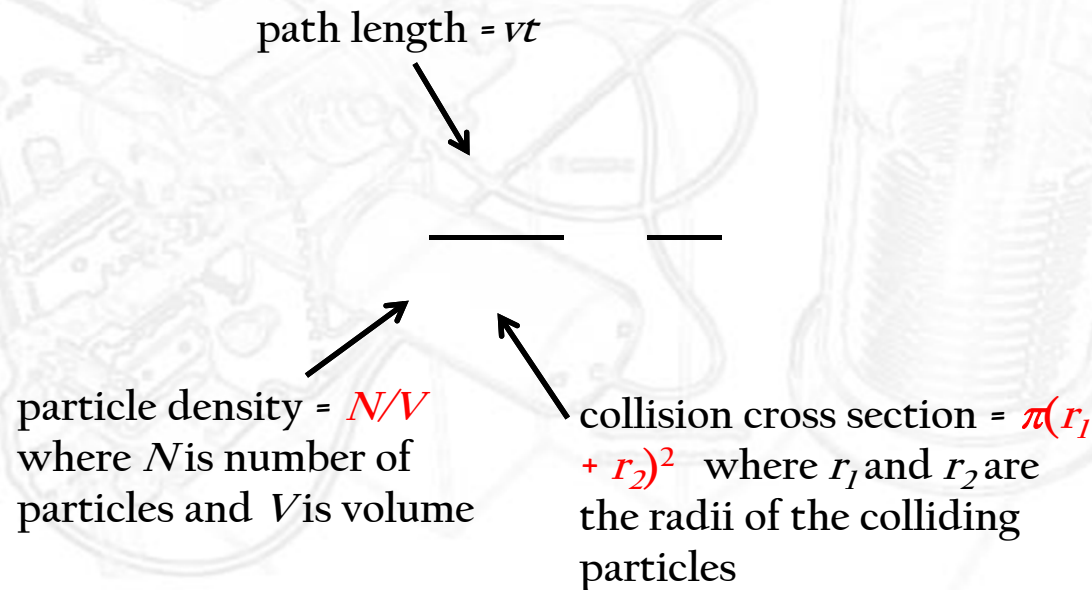
# Summing Up Mass Analyzers

Type	Resolution	Mass Accuracy	Sensitivity	MS/MS	Versatility
Sector				 	
Quadrupole					
Paul Trap				 	
TOF					
Orbitrap		 			
FT-ICR	 	 	 		

# Vacuum Systems: Who cares?

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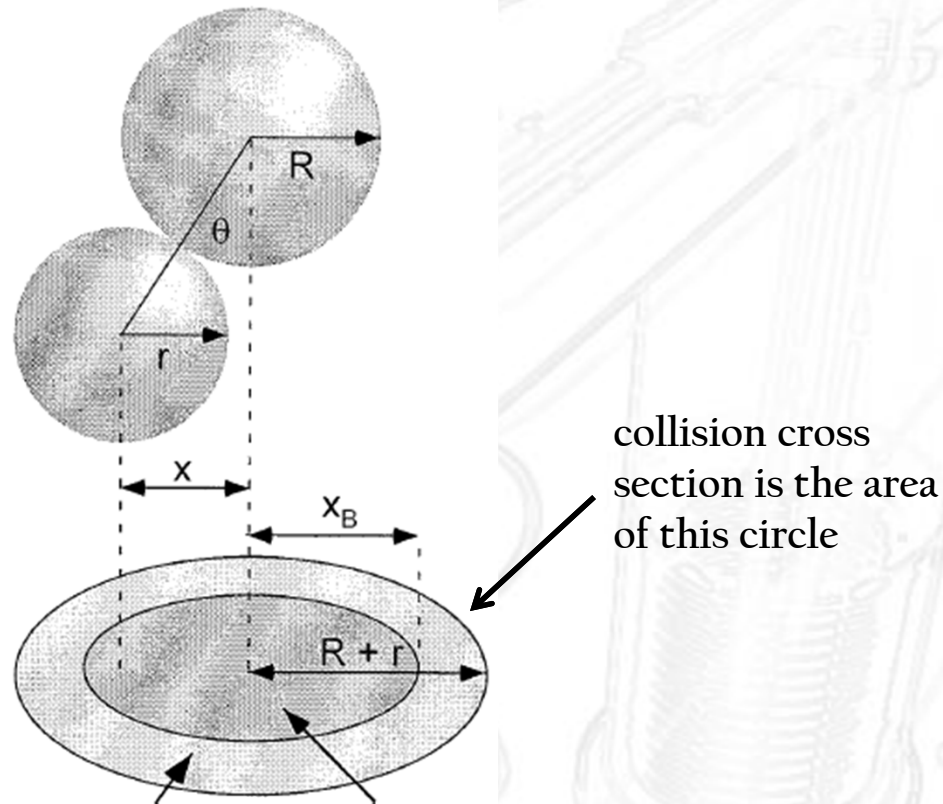
- Why do we care about vacuum systems? Only because **MS can't be done without them!**
- You may recall that the **first advances** in MS were due to the ability to make a **decent vacuum**.
- Why? It's all about the 'mean free path'  $\lambda$ :



# Collision Cross Section

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- The collision cross section is an important concept, particularly in ion mobility.



- Notice how we are assuming that the colliding particles are **spherical**...hmmmm....

# The Mean Free Path

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- The mean free path is the average distance that a particle will travel before colliding with another particle.
- Example: Lets say our favorite ion  $[\text{DEREK}+\text{H}]^+$  is flying through a **1 m** TOF tube (**radius 0.1 m**) at a pressure of  **$5 \times 10^{-8}$  Torr** of  **$\text{N}_2$**  at **room temp**.
- First, lets calculate the number of particles, N.

P (torr  $\rightarrow$  atm)

V ( $\text{m}^3 \rightarrow \text{L}$ )

$$\frac{(\quad)(\quad)}{(\quad)(\quad)}$$

## Mean Free Path Calc. Cont.

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- Now we can calculate the particle density:

$$\frac{1}{\lambda} = n \sigma$$

- How about the collision cross section:

$$\sigma = \pi d^2$$

- Finally, the mean free path:

$$\lambda = \frac{1}{n \sigma}$$

# Measuring Low Vacuum

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- Measurement of low vacuum is relatively easy because there is still plenty of gas around...
  - **Pirani** gauges simply measure resistance in a (usually platinum) wire exposed to vacuum. Current heats the wire up (resistance goes up), collisions with ambient gas cool the wire down (resistance goes down).
- Obviously no good if there isn't enough gas around to cause a measurable change in resistance.
- Pirani gauges are very accurate down to about  **$10^{-4}$  torr**.



# Measuring High Vacuums

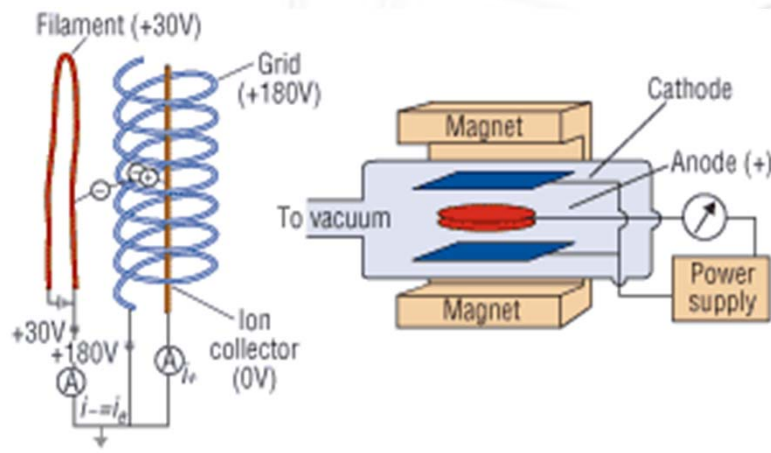
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- These days, MS instrument operate at ‘high vacuum’, i.e. anything lower than  $10^{-7}$  torr.
- The cheapest high vacuum measuring device is a **Penning** or **Cold Cathode** gauge.
- Electrons are generated via discharge in a cold cathode. Electrons flow to the anode and collide with gas particles, ionizing them. These ions are collected in an ‘ion collector’ (negatively charged cup).
- These gauges have upper pressure limit because if there is too much recombination, there is no current.
- On the other hand, if pressure too low, no discharge at cathode.
- Effective measurement range:  **$10^{-5} - 10^{-9}$  torr**



## Measuring High Vacuums Cont.

- The second cheapest, the **Bayard-Alpert** or **Hot Cathode** gauge works on a similar principle:



- Electrons generated from a heated filament ionize gas, which is drawn to a negatively charged filament inducing a measurable current. Electrons are discharged on a positively charged grid.

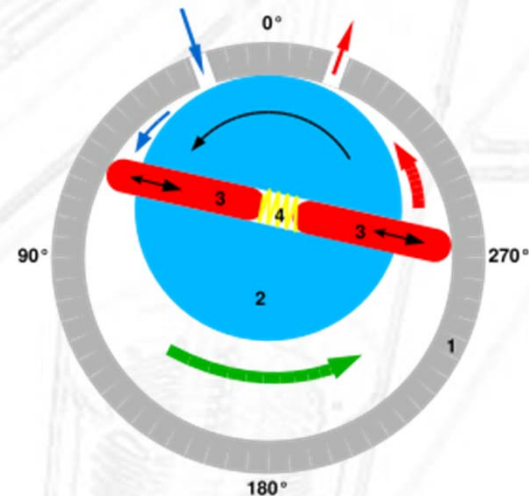
- Pressure range =  $10^{-3}$  to  $10^{-10}$  torr.



# Making a low Vacuum

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- Because of how we are going to make a high vacuum, we first need to make a low vacuum, maybe  $5 \times 10^{-4}$  torr or so...
- To do this, we use a **Rotary Vane** pump:



- These pumps have a slow pumping speed (about **3 L/min**) but can handle a lot of gas/volume (i.e. they can run at high pressure).
- They are also **exceedingly tough**.

# Making High Vacuum

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- Once we have a low / medium vacuum, we can use high flow pumping devices to make a high vacuum.
- The most common type is the **turbomolecular** pump, which is built like a jet engine:
- The mechanism is simple: Gas particles diffuse into the fan blades and are physically knocked away from the vacuum and into the pump.
- Successively flatter blades ensure that the particle keeps going in the right direction.



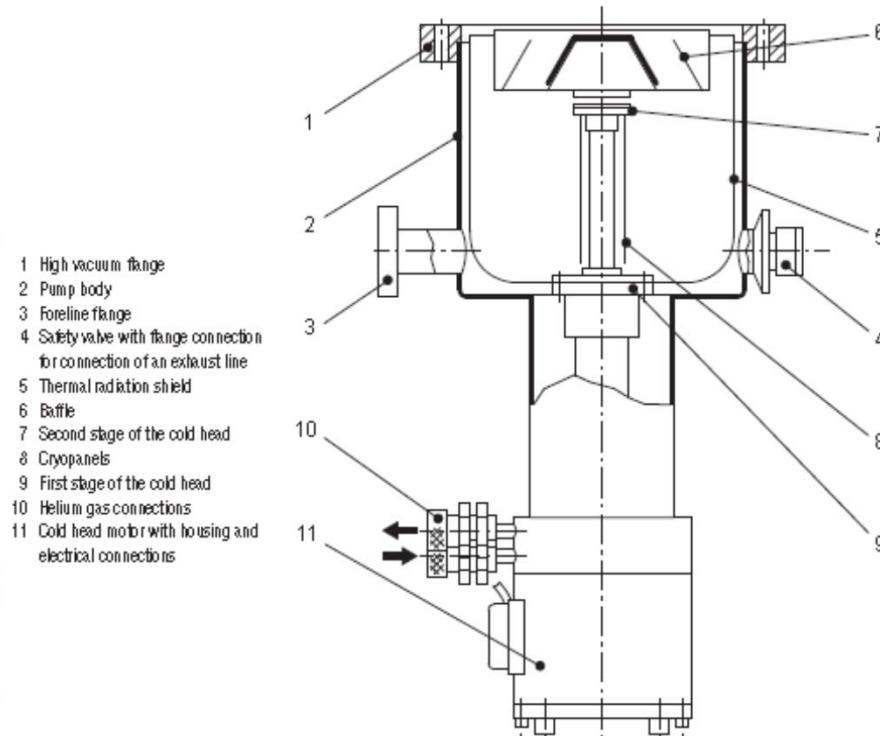
# Turbomolecular Pumps

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- The main advantage of turbopumps is that they require **relatively little power** (~ 100W) and can generate an oil-free vacuum.
- The main disadvantage is that they **are expensive** and **prone to sudden breakage** (not unlike a hard drive or anything else that spins).
- Pumping speed is around **300 L/sec**.
- The fan blades spin at 50 – 60,000 rpm. Obviously, we cannot do this at high pressure!

# Cryopumps

- Need a super-high vacuum? Try a cryopump:



- These pumps work by condensing gas onto a surface that is cooled to  $N_2$  temperatures.

# Cryopumps

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- Cryopumps cannot be used at high pressures because the adsorbent surface quickly becomes saturated. Thus they must be operated with a backing pump, or even a vane pump + turbomolecular system.
- In return, cryopumps offer extremely high pumping speeds, up to **1500 L/s**.
- This allows them to achieve the highest vacuum =  **$10^{-12}$  torr** or better if a colder cryogen is used.
- One disadvantage is that, even at high vacuum, the adsorbent eventually becomes saturated and must be replaced from time to time (which must be done out of vacuum).

# Vacuum Systems: Who cares?

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- Why do we care about vacuum systems? Only because **MS can't be done without them!**

